# Colorimetric Method for Determination of 7-Aminocephalosporanic Acid (7-ACA) and Related Compounds 

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#### Abstract

A reproducible colorimetric method is described for the determination of 7 -aminocephalosporanic acid (7-ACA) and related compounds.


ANUMBER OF ACYL derivatives of 7 -aminocephalosporanic acid ( $7-\mathrm{ACA}$ ) (I) and their antibacterial activities have been described (1-4). The D-phenylglycine derivative of 7-ACA, known generically as cephaloglycin ${ }^{1}$ (II), has a broad spectrum of antibacterial activity. To determine the extent of acylation in the production of cephaloglycin, an assay was needed for small concentrations of 7-ACA. Methodology for the assay of 6-aminopenicillanic acid (6-APA) (III) in penicillin derivatives was investigated; however, the methods encountered had two basic drawbacks: (a) specificity (separation techniques utilized) and (b) sensitivity.

This paper describes a chemical method which permits the direct determination of small amounts of 7-ACA ( $0.2-1.5 \%$ ) in cephaloglycin. At lower concentrations of $7-\mathrm{ACA}(0.04 \%)$, a $25 \%$ error may be incurred. Optimum conditions were established for the rapid development of the 7-ACAninhydrin chromophore. Essentially no color formation was obtained with either cephaloglycin or phenylglycine under the same conditions.
7-Aminodesacetoxycephalosporanic acid (7ADCA) (IV) and 6-aminopenicillanic acid (6-APA), both having an $\alpha$-amino group adjacent to a $\beta$ lactam ring in common with 7-ACA, were found to respond in a similar manner. Likewise, 7-ADCA could be determined directly in small amounts ( $0.4-1.5 \%$ ) in cephalexin ${ }^{2}(\mathrm{~V})$. At lower concentrations of 7-ADCA $(0.1 \%)$, a $25 \%$ error may be incurred.

## EXPERIMENTAL

Apparatus and Reagents-A spectrophotometer (Beckman model DU ) with $1-\mathrm{cm}$. glass cells was used in this laboratory, but any suitable spectrophotometer may be used. (a) Buffer solution- $2.0 \% \mathrm{w} / \mathrm{v}$ citric acid $\left(\mathrm{H}_{3} \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{O}_{7} \cdot \mathrm{H}_{2} \mathrm{O}\right)$ in 0.8 M sodium hydroxide; (b) formic acid-C.P., $98-100 \%$; (c) ninhydrin solution- $5.0 \% \mathrm{w} / \mathrm{v}$ ninhydrin (certified triketohydrindene hydrate, Fisher Scientific Co.) in methylcellosolve (peroxide-free).

Recommended Procedure-Method I-Determination of 7-ACA-Accurately weigh approximately 7 mg . ( $\pm 0.01 \mathrm{mg}$.) of sample and $7-\mathrm{ACA}$ reference standard, and quantitatively transfer each to a $100-\mathrm{ml}$. volumetric flask. Dissolve the contents in 5.0 ml . of formic acid, dilute to volume with distilled

[^0]


I




IV


V
$\mathrm{H}_{2} \mathrm{O}$, and mix thoroughly. Prepare a reference blank solution in a $100-\mathrm{ml}$. volumetric flask by diluting 5.0 ml . of formic acid to volume with distilled $\mathrm{H}_{2} \mathrm{O}$ and mixing well. Add by transfer pipet to a test tube in the following order: 2.0 ml . of each prepared solution, 1.0 ml . of buffer solution, and 0.50 ml . of ninhydrin solution. Mix thoroughly, and after a $9-\mathrm{min}$. waiting period, determine the absorbance of each solution at $407 \mathrm{~m} \mu$ on a suitable spectrophotometer, using $1.0-\mathrm{cm}$. glass cells and the reference blank solution in the reference cell.

Calculations:
abs. 407 sample $\times \mathrm{mg}$. standard
$\stackrel{\text { abs. } 407 \text { standard } \times \text { mg. sample }}{\text { sta }} \times 100=\%$-ACA
Method II-7-ACA in Cephaloglycin-Accurately weigh approximately 100 mg . ( $\pm 0.1 \mathrm{mg}$.) of cephaloglycin sample, and quantitatively transfer to a $10-\mathrm{ml}$. volumetric flask. Dissolve the contents in 0.50 ml . of formic acid, dilute to volume with distilled $\mathrm{H}_{2} \mathrm{O}$, and mix thoroughly. Prepare a reference standard stock solution by dissolving 70 mg . of 7-ACA reference standard in 50 ml . of formic acid


Fig. 1-Rate of color formation of 7-ACA chromophore.
and mixing well. Add 0.50 ml . of this stock solution to a $10-\mathrm{ml}$. volumetric flask, dilute to volume with distilled $\mathrm{H}_{2} \mathrm{O}$, and mix thoroughly. Prepare a reference blank solution in a $10-\mathrm{ml}$. volumetric flask by diluting 0.50 ml . of formic acid to volume with distilled $\mathrm{H}_{2} \mathrm{O}$ and mixing well. Continue the assay as outlined in Method I, and calculate accordingly.

## DISCUSSION

Standard Curve-A plot of the absorbance versus concentration was found to be linear and to pass through the origin, obeying Beer's law from 4-40 meg./ml.
Stability of the Color-The color formation increases uniformly and rapidly, reaching a maximum in 9 min ., then decreasing (Fig. 1). The yellow chromophore described exhibits a peak at $407 \mathrm{~m} \mu$; a secondary peak is also present at $570 \mathrm{~m} \mu$ (Fig. 2).
Influence of Reagents-The color intensity was found to increase with a corresponding increase in pH , reaching a maximum at pH 4.5 , then decreasing. An increase in pH not only influenced the rate of color development but also shifted the maximum absorbance peak to a higher wavelength ( $415 \mathrm{~m} \mu$ ). The intensity of the secondary peak ( $570 \mathrm{~m} \mu$ ) decreased with a pH increase. A pH 3 buffered system was selected, since cephaloglycin was found to be somewhat reactive at a higher pH . Both of the final concentrations of ninhydrin and methylcellosolve were found to influence the color formation. Optimum color formation was obtained when the final concentration of methylcellosolve (containing $50 \mathrm{mg} . / \mathrm{ml}$. of ninhydrin) was set at $14.3 \%$ $\mathbf{v} / \mathbf{v}$. Temperature fluctuations of $\pm 3^{\circ}$, as well as light, did not affect the reaction rate. The effect of the 7-ACA standing in formic acid was found to be insignificant ( $0-3 \mathrm{~min}$.).
Interferences-Several compounds were assayed


Fig. 2-Cary recording spectrophotometric curves of chromophores.

Table I-Absorptivity (a) at $407 \mathrm{~m} \mu$

| Compd. | $a$ |
| :--- | :---: |
| 7-ACA | 15.3 |
| 7-ADCA | 11.0 |
| 6-APA | 5.4 |
| Cephaloglycin | 0.0018 |
| Cephalexin | 0.0035 |
| Phenylglycine | 0.33 |
| $\mathrm{NH}_{4} \mathrm{Cl}$ | 0.16 |

by this procedure to determine the extent of color formation, as shown in Table I.

Reproducibility-The following standard deviations were obtained on the methods outlined in Methods I and II of the Procedure: Method I$2 s= \pm 3.7 \%, n=6 ;$ Method $\mathrm{II}-2 s= \pm 4.2 \%$, $n=6$.

A sample of cephaloglycin was assayed for 7-ACA by the colorimetric method, as well as by the amino acid analyzer. Agreement was obtained as follows: amino acid analyzer- $1.35 \%$; colorimetric method $-1.42 \%$.

7-ADCA and 6-APA-Essentially the same procedure was found to produce characteristic chromophores with 6-APA and 7-ADCA (Fig. 2); however, the rates of formation differed. In addition, the 7-ADCA color curve exhibited a third peak at 480 $m \mu$ which proved to be more suited to the direct determination of 7-ADCA in cephalexin. Maximum color development for the 7-ADCA chromophore was obtained in 18 min . at $407 \mathrm{~m} \mu$ and in 50 min . at $480 \mathrm{~m} \mu$. The 6-APA chromophore reached maximum color development in 45 min . at $407 \mathrm{~m} \mu$.

The absorptivity for Compound 7-ADCA was 12.2 at $407 \mathrm{~m} \mu$ and 11.5 at $480 \mathrm{~m} \mu$; for Compound cephalexin $a=0.0065$ at $407 \mathrm{~m} \mu$ and $a=0.0035$ at $480 \mathrm{~m} \mu$; for Compound 6-APA $a=9.55$ at $407 \mathrm{~m} \mu$.

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7-Aminocephalosporanic acid ( 7 -ACA), related compounds-analysis Cephaloglycin-7-ACA determination Ninhydrin-color reagent
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    ${ }^{1}$ Cephaloglycin is the generic name given to 7 -( $\mathrm{D}-\alpha-$ aminophenylacetamido) cephalosporanic acid.
    ${ }^{2}$ Cephalexin is the generic name given to 7 - ( $\mathrm{d}-\alpha$-aminophenylacetamido) desacetoxy cephalosporanic acid.

